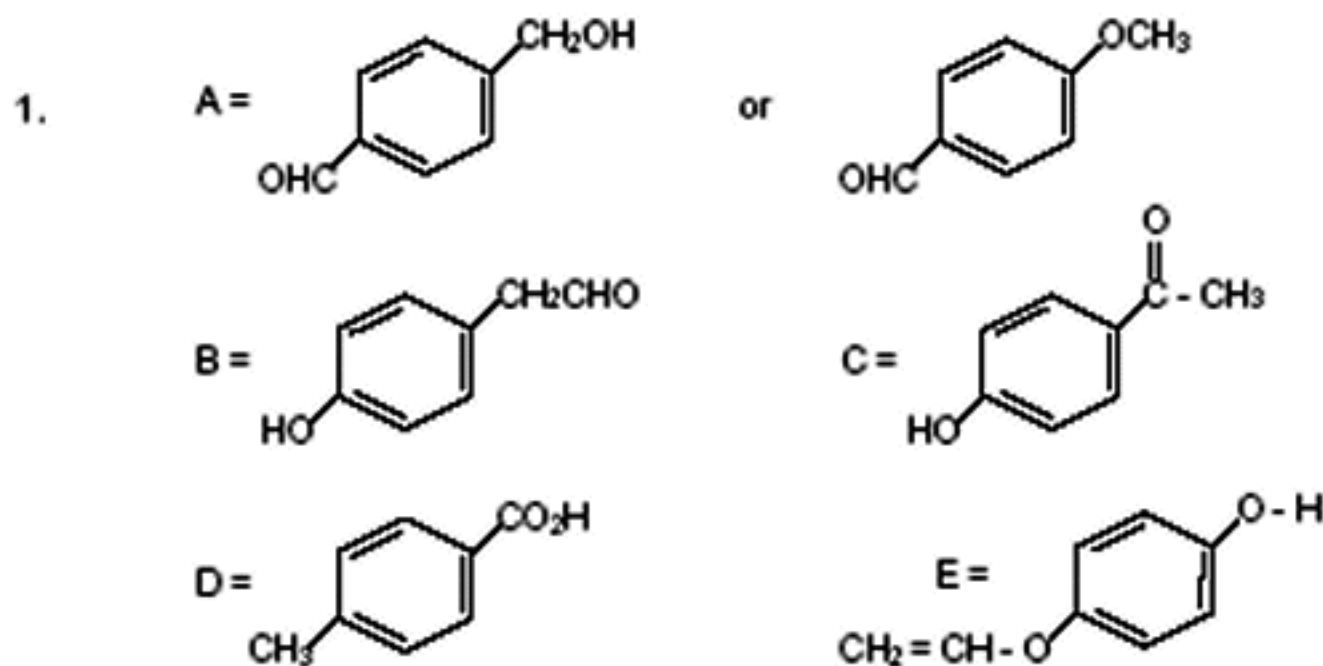


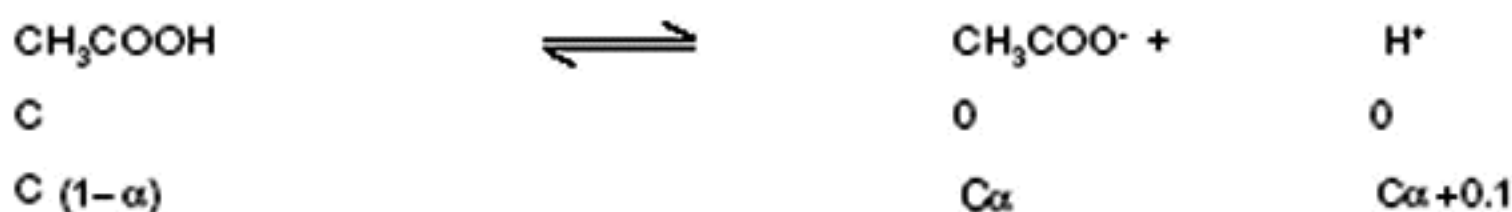
Solution Chemistry



2. (i) The volume being doubled by mixing the two solutions, the molarity of each component will be halved i.e.

$$[\text{CH}_3\text{COOH}] = 0.1 \text{ M}, [\text{HCl}] = 0.1 \text{ M}$$

HCl being a strong acid will remain completely ionised and hence H^+ ion concentration furnished by it will be 0.1 M. This would exert common ion effect on the dissociation of acetic acid, a weak acid.



$$K_a = \frac{C\alpha\alpha(C+0.1)}{C(1-\alpha)} = \frac{C\alpha^2 + 0.1\alpha}{(1-\alpha)}$$

Neglecting α in comparison to unity and $C\alpha^2$ i.e. $0.1\alpha^2$ in comparison to 0.1α , we get $K_a = 0.1\alpha$

$$\text{or } \alpha = \frac{K_a}{0.1} = \frac{1.75 \times 10^{-5}}{0.1} = 1.75 \times 10^{-4}$$

$$[\text{H}^+]_{\text{Total}} = 0.1 + C\alpha, C\alpha \text{ is negligible as compared to } 0.1$$

$$\therefore [\text{H}^+]_{\text{Total}} = 0.1$$

$$\therefore \text{pH} = 1$$

$$(ii) 6 \text{ g NaOH} = \frac{6}{40} = 0.15 \text{ mol}$$

0.1 mole of NaOH will be consumed by 0.1 mole of HCl. Thus, 0.05 mole of NaOH will react with acetic acid (No. of mol of $\text{CH}_3\text{COOH} = 1 \times 0.1 = 0.1$) according to the equation.



Thus, solution of acetic acid and sodium acetate will become acidic buffer. So pH of the buffer will be

$$\text{pH} = \text{p}K_a + \log \frac{[\text{salt}]}{[\text{acid}]} = -\log(1.75 \times 10^{-5}) + \log 1 = 4.75$$

3. $[\text{NiCl}_4]^{2-} \Rightarrow sp^3$ (as Cl^- is weak field ligand) - Tetrahedral
 $[\text{Ni}(\text{CN})_4]^{2-} \Rightarrow dsp^2$ (as CN^- is strong field ligand) - Square planar

Magnetic moments (μ_{spin}) values are as follows,

$$[\text{NiCl}_4]^{2-} \Rightarrow \sqrt{2(2+2)} = 2.82 \text{ BM}$$

4. a) $d = 0.36 \text{ kg m}^{-3} = 0.36 \text{ g/L}$
 (i) From Graham's Law of diffusion

$$\frac{r_v}{r_{\text{O}_2}} = \sqrt{\frac{M_{\text{O}_2}}{M_v}}; 1.33 = \sqrt{\frac{32}{M_v}} \therefore M_v = \frac{32}{(1.33)^2} = 18.09, \text{ where } M_v = \text{MW of the vapour}$$

$$(ii) \text{ Thus, } 0.36 \text{ g} = \frac{0.36}{18.09} \text{ mol}$$

$$\frac{0.36}{18.09} \text{ mol mol occupies 1 L volume}$$

$$\text{so 1 mol occupies } \frac{18.09}{0.36} \text{ L}$$

Thus, molar volume of vapour = 50.25 L

Assuming ideal behaviour the volume of the vapour would be

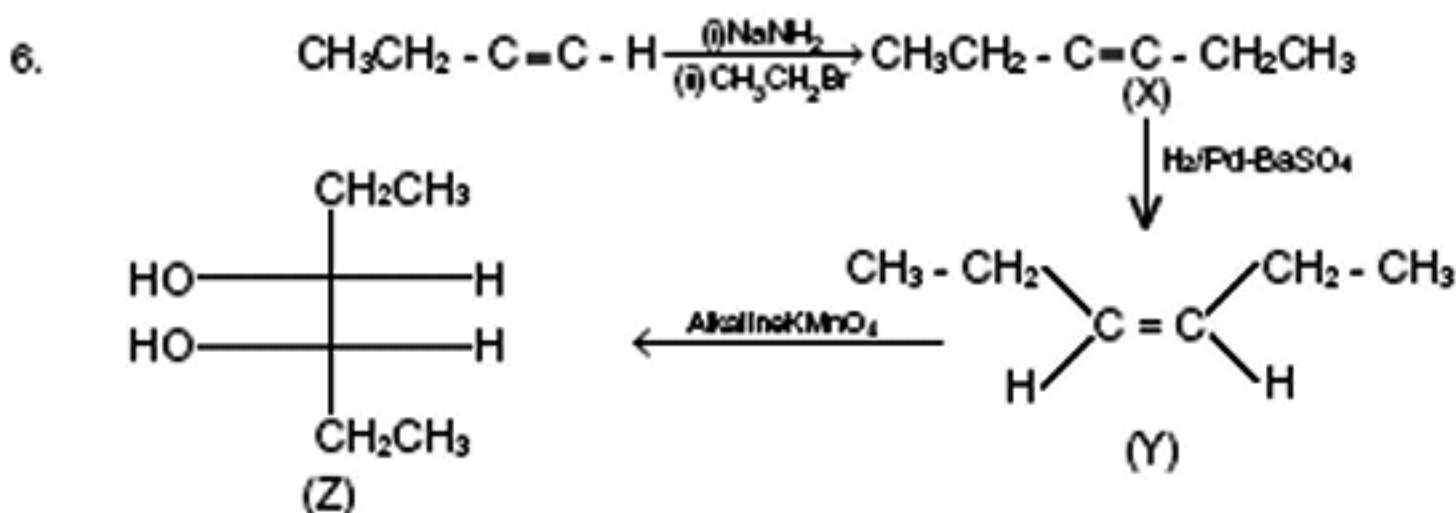
$$\frac{V_1}{T_1} = \frac{V_2}{T_2} \Rightarrow V_2 = 22.4 \times \frac{500}{273} = 41.025 \text{ L}$$

$$(ii) \text{ Compressibility factor } (Z) = \frac{(PV)_{\text{obs}}}{(PV)_{\text{ideal}}} = \frac{1 \times 50.25}{1 \times 41.025} = 1.224$$

(iv) Z being greater than unity, it is the short range repulsive force that would dominate.

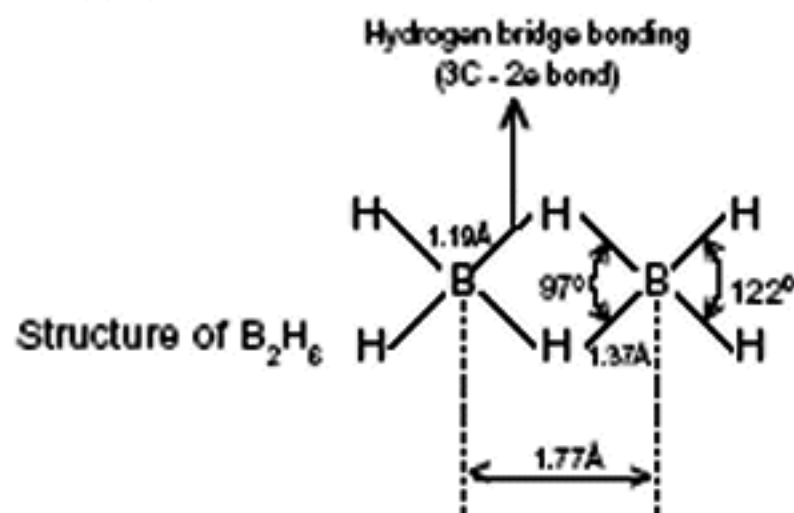
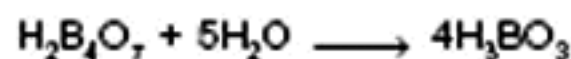
$$(b) E = \frac{3}{2} k_b T = \frac{3}{2} \times 1.38 \times 10^{-23} \times 1000 = 2.07 \times 10^{-20} \text{ J per molecule.}$$

5. (I) $\text{Al}_4\text{C}_3 + 12\text{H}_2\text{O} \longrightarrow 4\text{Al}(\text{OH})_3 + 3\text{CH}_4 \uparrow$
 (II) $\text{CaNCN} + 3\text{H}_2\text{O} \longrightarrow \text{CaCO}_3 \downarrow + 2\text{NH}_3$
 Ammonia formed dissolves in water to form NH_4OH
 $\text{CaNCN} + 5\text{H}_2\text{O} \longrightarrow 2\text{NH}_4\text{OH} + \text{CaCO}_3 \downarrow$
 (III) $4\text{BF}_3 + 3\text{H}_2\text{O} \longrightarrow 3\text{HBF}_4 + \text{B}(\text{OH})_3$ (IV) $\text{NCl}_3 + 3\text{H}_2\text{O} \longrightarrow \text{NH}_3 + 3\text{HOCl}$
 (V) $3\text{XeF}_4 + 6\text{H}_2\text{O} \longrightarrow \text{XeO}_3 + 2\text{Xe} + 3/2\text{O}_2 + 12\text{HF}$

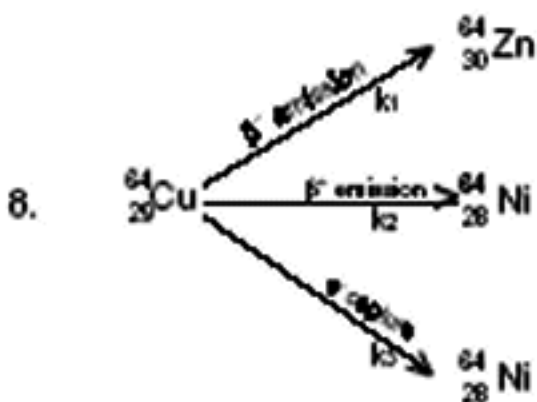


Z is in meso form having plane of symmetry. The upper half molecule is mirror image to the lower half molecule. The molecule is, therefore, optically inactive due to internal compensation.

7. When hot concentrated HCl is added to borax ($\text{Na}_2\text{B}_4\text{O}_7 \cdot 10\text{H}_2\text{O}$) the sparingly soluble H_3BO_3 is formed which on subsequent heating gives B_2O_3 which is reduced to boron on heating with Mg, Na or K.



Normally this reaction takes place in the presence of Lewis acid (AlCl_3)



Let the rate constants of the above emission processes be k_1 , k_2 and k_3 , respectively and the overall rate constant be k . Then

$$k = k_1 + k_2 + k_3 = \frac{0.693}{t_{1/2}} = \frac{0.693}{12.8} \text{h}^{-1}$$

$$\text{Also, } k_1 = 0.38k = 0.38 \times \frac{0.693}{12.8} \text{h}^{-1}$$

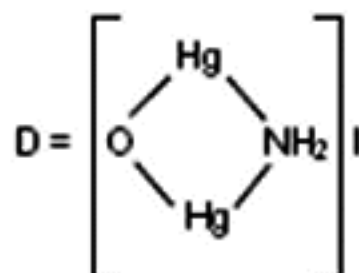
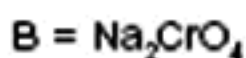
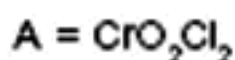
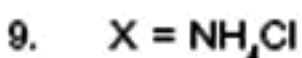
$$t_1 = \frac{0.693 \times 12.8}{0.38 \times 0.693} = 33.68 \text{h}$$

Similarly,

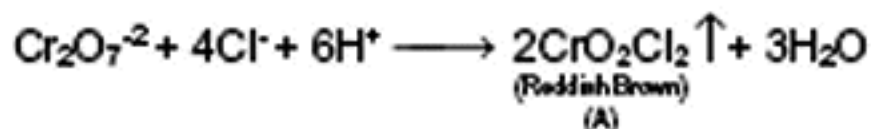
$$t_2 = \frac{0.693}{k_2} = \frac{0.693}{0.19k} = \frac{0.693}{0.19 \times 0.693} \times 12.8 = 67.36 \text{h}$$

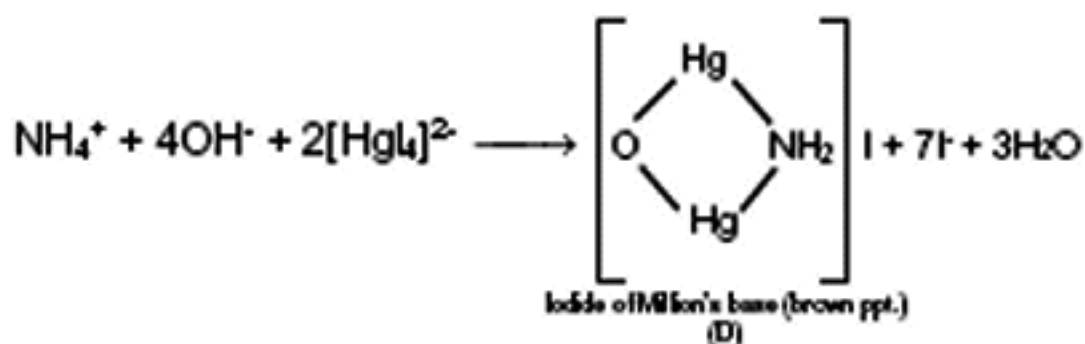
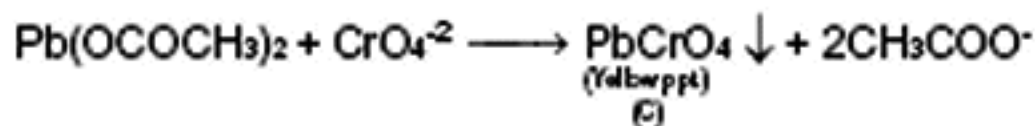
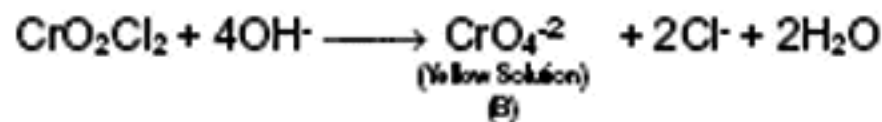
$$t_3 = \frac{0.693}{k_3} = \frac{0.693}{0.43k} = \frac{12.8}{0.43} = 29.76 \text{h}$$

where t_1 , t_2 and t_3 are the partial half lives for β^- emission, β^+ emission and electron capture processes, respectively.

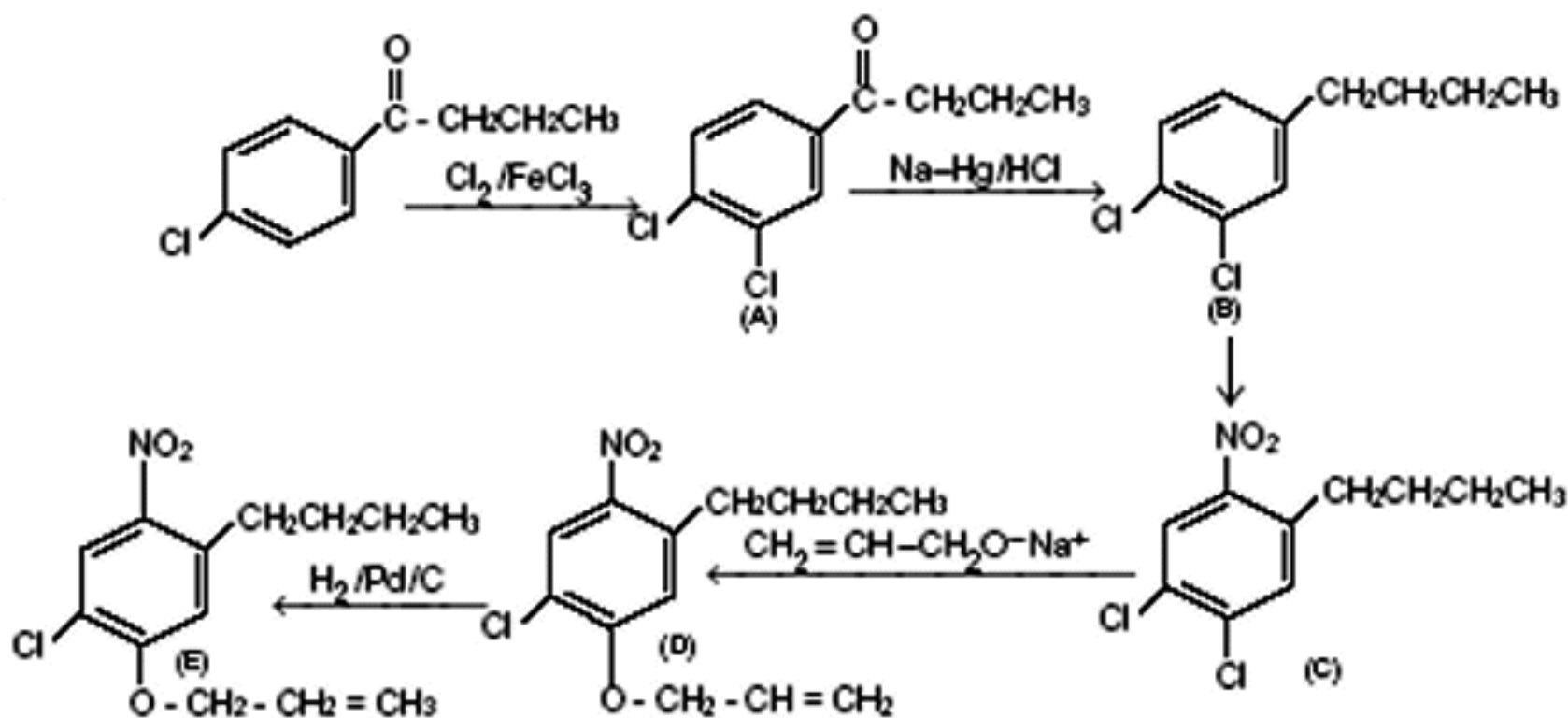


The chemical reactions are as follows:

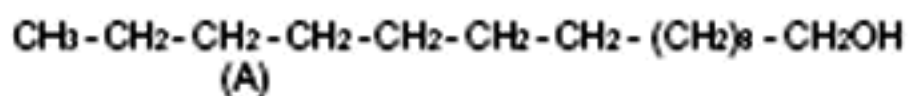
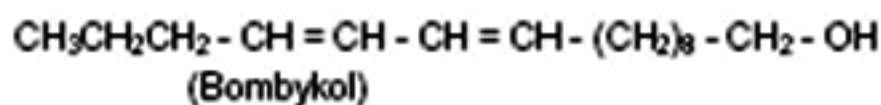




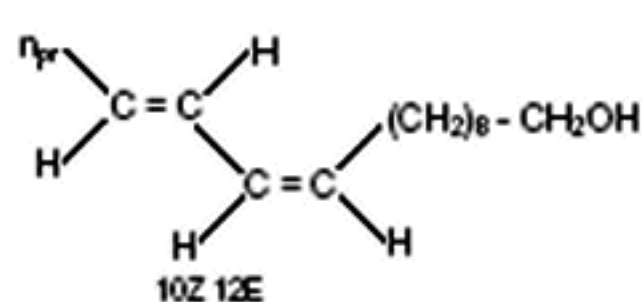
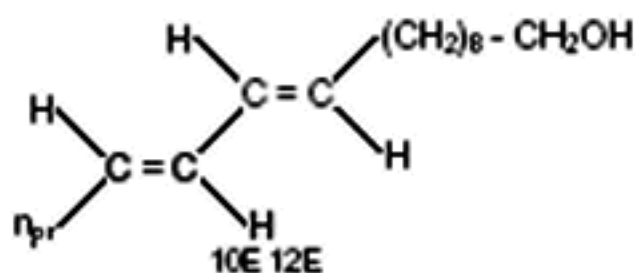
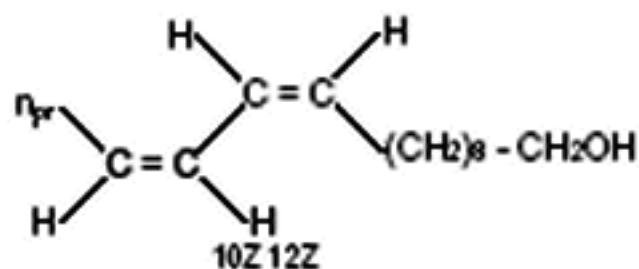
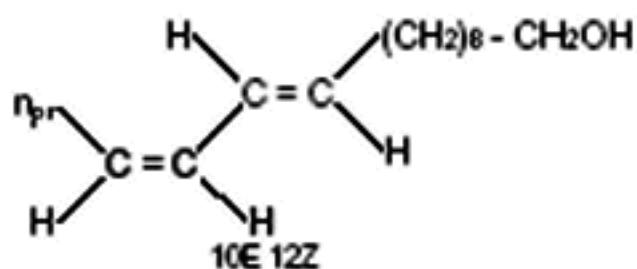
10.



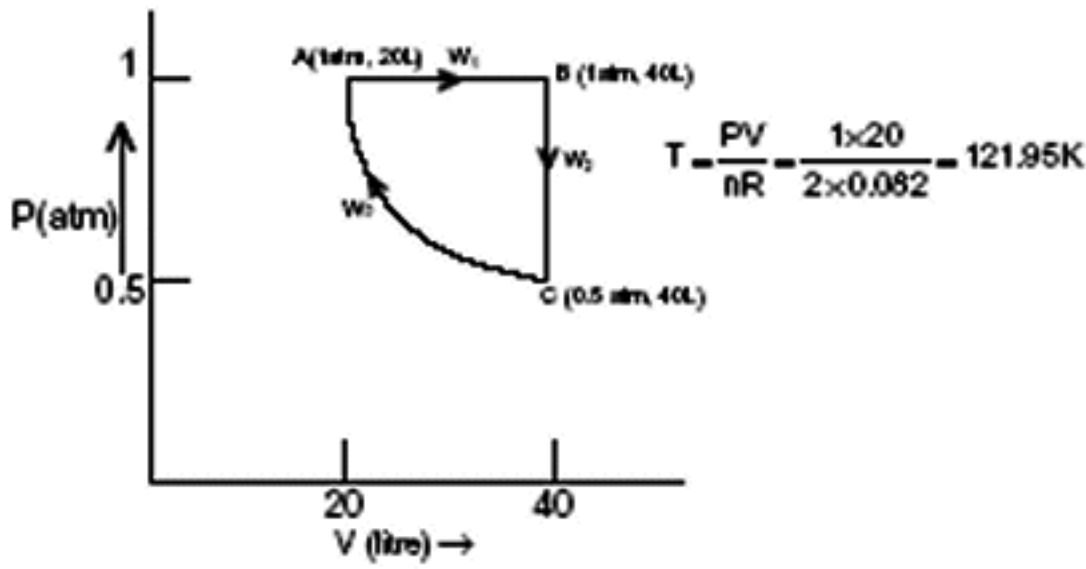
11. No. of double bonds = 2



4 Geometrical Isomers are possible.



12. (i)



(ii) Total work (W) = $W_1 + W_2 + W_3$

$$= -P\Delta V + 0 + 2.303 nRT \log \frac{V_1}{V_2}$$

$$= -1 \times 20 + 2.303 \times 2 \times 0.082 \times 121.95 \log 2$$

$$= -20 + 13.86 = -6.13 \text{ L atm}$$

Since the system has returned to its initial state i.e. the process is cyclic so $\Delta U = 0$

$$\Delta U = q + W = 0, \text{ so}$$

$$q = -W = 6.13 \text{ L atm} = 620.7 \text{ J}$$

In a cyclic process heat absorbed is completely converted into work

(iii) Entropy is a state function and since the system has returned to its initial state so $\Delta S = 0$.

Similarly $\Delta H = 0$ and $\Delta U = 0$ for the same reason i.e. U and H are also state functions having definite values in a given state of a system.

Solution Physics

1. (a) $2(\text{Fundamental Frequency of A}) = 3(\text{Fundamental Frequency of B})$

$$\frac{2v_A}{2l_A} = \frac{3v_B}{4l_B}$$

As, $v = \sqrt{\frac{\gamma RT}{M}}$, we have $v = \sqrt{\frac{\gamma_A}{M_A}} = \frac{3}{4} \sqrt{\frac{\gamma_B}{M_B}}$; $\frac{M_A}{M_B} = \frac{400}{180}$

(b) $\frac{v_A}{v_B} = \sqrt{\frac{\gamma_A M_B}{\gamma_B M_A}} = \frac{3}{4}$

2. (a) Time between two consecutive collisions = $\frac{2l}{v} = t$

Here, $t = \frac{1}{500}$ sec and $l = 1\text{m} \Rightarrow v = 1000 \text{ m/s}$. Also, $v = \sqrt{\frac{3RT}{M}} \Rightarrow T = 160\text{K}$

- (b) Average kinetic energy of an atom of a monoatomic gas = $\frac{3}{2}kT$

$$\therefore E_{av} = \frac{3}{2}kT = 3.312 \times 10^{-21} \text{ Joules}$$

- (c) From gas equation $PV = \frac{m}{M}RT \Rightarrow m = 0.3012\text{gm}$

3. (a) As the pressure force exerted by liquid A is radial & symmetric its net value is zero.

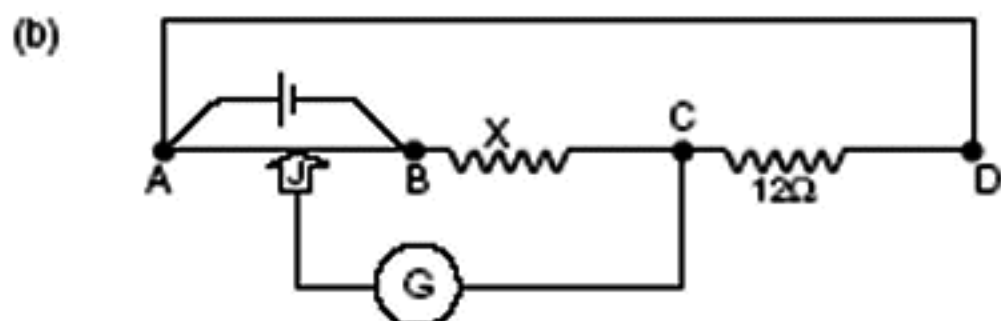
- (b) For equilibrium, Buoyant force = weight of the body

$$\Rightarrow h_A \rho_A Ag + h_B \rho_B Ag = (h_A + h + h_B) A \rho_c g \quad (\text{where } \rho_c = \text{density of cylinder})$$

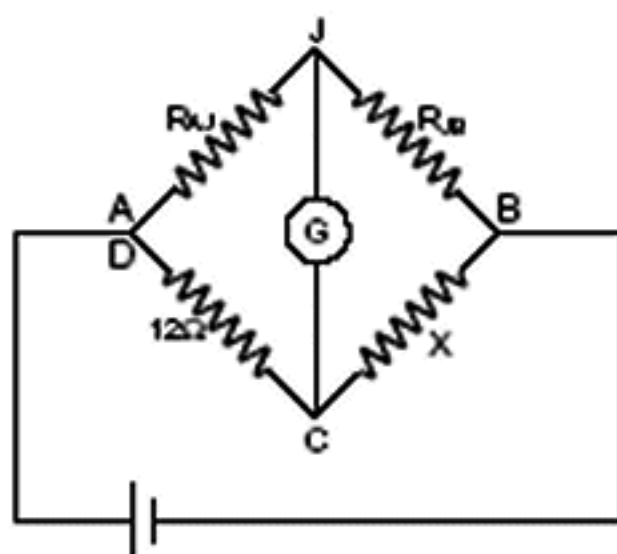
$$h = \left(\frac{h_A \rho_A + h_B \rho_B}{\rho_c} \right) - (h_A + h_B) = 0.25\text{cm}$$

- (c) $a = \frac{F'_{\text{Buoyant}} - Mg}{M} = \left[\frac{h_A \rho_A + \rho_B (h + h_B) - (h + h_A + h_B) \rho_c}{\rho_c (h + h_A + h_B)} \right] g = \frac{g}{6}$ upwards

4. (a) No



(c) ∴ Bridge is balanced



$$\frac{R_{AJ}}{R_{CB}} = \frac{0.6\rho}{0.4\rho} = \frac{12\Omega}{X} \Rightarrow X = 8\Omega$$

Where ρ is resistance per unit length.

5. (a) If x is the difference in quantum number of two states then ${}^xP_2 = 6 \Rightarrow x = 3$

$$\text{Now, we have } \frac{-z^2(13.6\text{eV})}{n^2} = -0.85\text{eV} \quad \dots\dots(1)$$

$$\text{and } \frac{-z^2(13.6\text{eV})}{(n+3)^2} = -0.544\text{eV} \quad \dots\dots(2)$$

Solving (1) and (2) we get $n = 12$ and $z = 3$

(b) Smallest wavelength λ is given by $\frac{hc}{\lambda} = (0.85 - 0.544) \text{ eV}$

Solving, we get $\lambda = 4052\text{nm}$.

6. (a) As there is symmetry about the line SP, fringes will be circular.

$$(b) \frac{I_{\min}}{I_{\max}} = \left(\frac{\sqrt{1} - \sqrt{0.36}}{\sqrt{1} + \sqrt{0.36}} \right)^2 = \left(\frac{0.4}{1.6} \right)^2 = \frac{1}{16}$$

(c) For maximum at P, path difference = $n\lambda$.

If AB is shifted by a distance x , it will cause an additional path difference of $2x$.

$$2x = \lambda \text{ (for minimum value of } x) \Rightarrow x = \lambda/2 = 300\text{nm}$$

7. (a) Torque due to magnetic forces should act opposite to that of gravity i.e. along the -ve y-axis. If $M\hat{k}$ is the magnetic moment

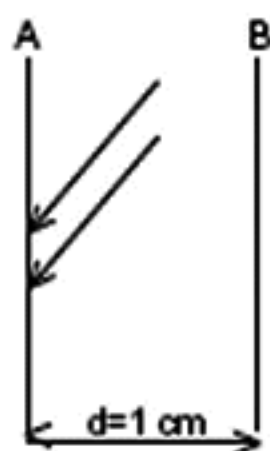
$$\vec{\tau}_b = \vec{M} \times \vec{B} = M\hat{k} \times (3\hat{i} + 4\hat{k})B_0 = 3MB_0\hat{j} \Rightarrow M \text{ is -ve}$$

\therefore I should be clockwise i.e. from P to Q

$$(b) \vec{F} = I(\vec{L} \times \vec{B}) \Rightarrow \vec{F}_{\text{res}} = I[(-b\hat{j}) \times (3\hat{i} + 4\hat{k})B_0] = IB_0b[3\hat{k} - 4\hat{i}]$$

$$(c) 3(ab)B_0 = mga/2 \Rightarrow I = \frac{mg}{6B_0b}$$

8. (a) Number of photoelectrons emitted from plate A upto $t = 10$ s



$$n_e = \frac{(5 \times 10^{-4}) \times 10^{16}}{10^6} \times 10 = 5 \times 10^7$$

(b) Charge on plate B at $t = 10$ sec

$$Q_b = 33.7 \times 10^{-12} - 5 \times 10^7 \times 1.6 \times 10^{-19} = 25.7 \times 10^{-12} \text{ C}$$

also $Q_a \times 10^{-12} \text{ C}$

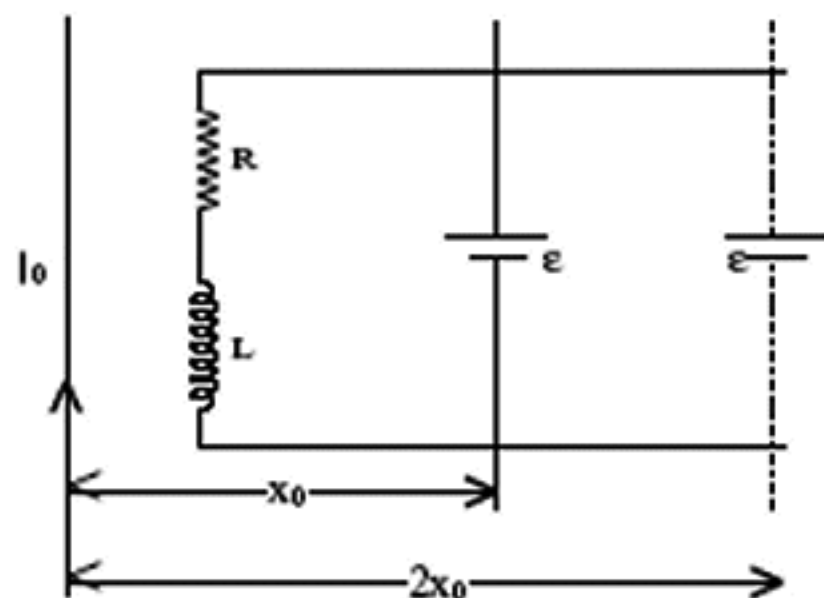
$$E = \frac{\sigma_b}{2\epsilon_0} - \frac{\sigma_a}{2\epsilon_0} = \frac{1}{2A\epsilon_0} (Q_b - Q_a)$$

$$= \frac{17.7 \times 10^{-12}}{5 \times 10^{-4} \times 8.85 \times 10^{-12}} = 2000 \text{ N/C}$$

(c) K.E. of most energetic particles = $(h\nu - \phi) + e(Ed) = 23\text{eV}$

9. (a) $\epsilon - L \frac{di}{dt} - iR = 0 \Rightarrow \left| \frac{d\phi}{dt} \right| - \frac{L di}{dt} = iR$

(b) Net change in flux = $\int_{x_0}^{2x_0} \frac{\mu_0 I_0}{2\pi x} dx - Li_1 = \frac{\mu_0 I_0}{2\pi} \ln 2 - Li_1$



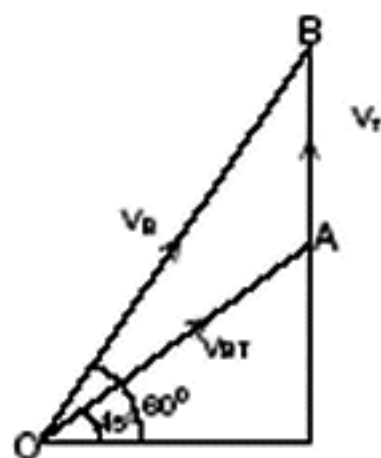
Net charge flown through resistance $R = \frac{\text{total change in flux}}{R} = \frac{\frac{\mu_0 I_0}{2\pi} \ln 2 - Li_1}{R}$

(c) Current in the circuit for $T \leq t \leq 2T$ is given by $I = I_1 e^{-t \cdot \eta R / L}$

At $t = 2T$, $I = I_1/4 \Rightarrow \frac{L}{R} = \frac{2T}{\ln 2}$

10. (a) From the diagram \vec{V}_{BT} makes an angle of 45° with the x-axis.

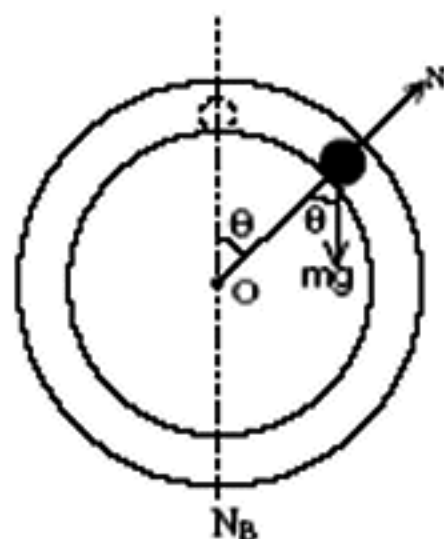
(b) Using sine rule



$$\frac{V_B}{\sin 135^\circ} = \frac{V_r}{\sin 15^\circ}$$

$$\Rightarrow V_B = 2 \text{ m/s}$$

11. (a) From F.B.D. of the ball



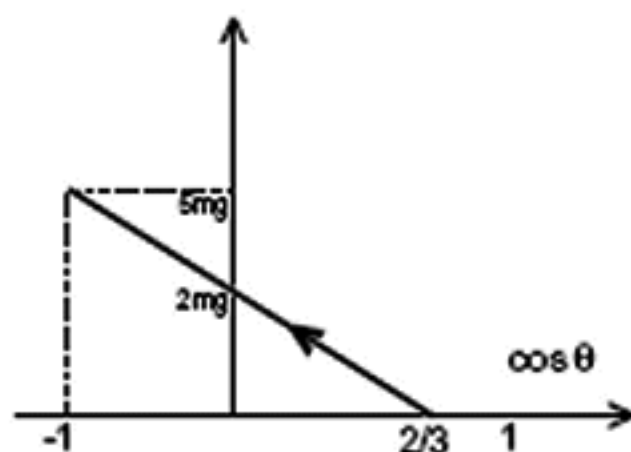
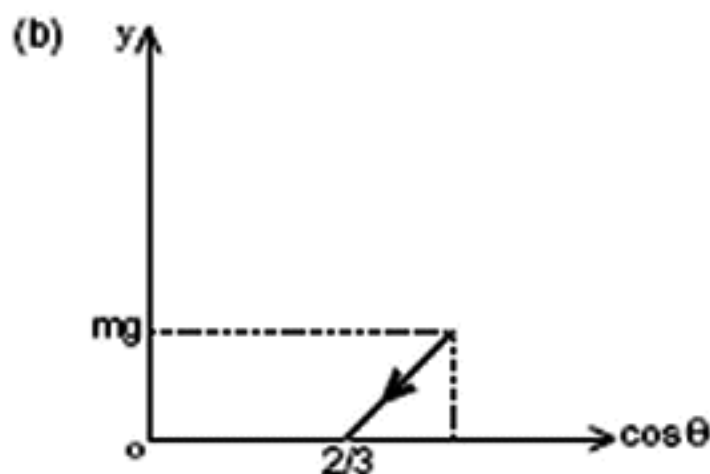
$$mg \cos \theta - N = \frac{mv^2}{(R + d/2)} \quad \dots\dots\dots (I)$$

Using conservation of energy

$$mg[R + (d/2)] (1 - \cos \theta) = (\frac{1}{2}) mv^2 \quad \dots\dots\dots (II)$$

From (I) and (II)

$$N = mg (3 \cos \theta - 2) \text{ radially outwards}$$



12. (a) Magnitude of centrifugal force on both B and C is

$$F_1 = m\omega^2 l$$

∴ Resultant force

$$F = 2F_1 \cos 30^\circ = \sqrt{3} m\omega^2 l$$

(b) Torque due to F about A

$$\tau_A = \frac{F\sqrt{3}l}{2} = I\alpha = 2ml^2\alpha$$

$$a_{\text{centrifugal}} = \alpha \frac{l}{\sqrt{3}} \Rightarrow F_x = -\frac{F}{4m}$$

$$F_x + F = \frac{F}{4m} 3m = \frac{3F}{4} \Rightarrow F_x = -\frac{F}{4}$$

$$F_y = \sqrt{3} ml\omega^2$$

Solution Mathematics

1. Clearly $A_1 + A_2 = a + b$

$$\frac{1}{H_1} + \frac{1}{H_2} = \frac{1}{a} + \frac{1}{b} \Rightarrow \frac{H_1 + H_2}{H_1 H_2} = \frac{a + b}{ab} = \frac{A_1 + A_2}{G_1 G_2}$$

$$\Rightarrow \frac{G_1 G_2}{H_1 H_2} = \frac{A_1 + A_2}{H_1 + H_2}$$

$$\text{Also } \frac{1}{H_1} = \frac{1}{a} + \frac{1}{3} \left(\frac{1}{b} - \frac{1}{a} \right) \Rightarrow H_1 = \frac{3ab}{2a + b}$$

$$\text{and } \frac{1}{H_2} = \frac{1}{a} + \frac{2}{3} \left(\frac{1}{b} - \frac{1}{a} \right) \Rightarrow H_2 = \frac{3ab}{2a + a}$$

$$\Rightarrow \frac{A_1 + A_2}{H_1 + H_2} = \frac{a + b}{3ab \left(\frac{1}{2b + a} + \frac{1}{2a + b} \right)} = \frac{(2b + a)(2a + b)}{9ab}$$

2. Let $P(n) : (25)^{n+1} - 24n + 5735$

$$\text{For } n = 1, P(1) : 625 - 24 + 5735 = 6336 = (24)^2 \times (11)$$

which is divisible by 24^2

Hence $P(1)$ is true

Let $P(k)$ be the true, where $k \geq 1$

$$\Rightarrow (25)^{k+1} - 24k + 5735 = (24)^2 \lambda \quad \text{where } \lambda \in \mathbb{N}$$

For $n = k + 1$

$$P(k+1) : (25)^{k+2} - 24k + 5735$$

$$= 25[(25)^{k+1} - 24k + 5735] + 25 \cdot 24 \cdot k - (25)(5735) + 5735 - 24(k+1)$$

$$= 25((24)^2 \lambda + (24)^2 k - 5735 \times 24 - 24)$$

$$= 25((24)^2 \lambda + (24)^2 k - (24)(5736))$$

$$= 25((24)^2 \lambda + (24)^2 k - (24)^2 (239))$$

$$= (24)^2 [25\lambda + k - 239]$$

which is divisible by $(24)^2$

Hence, by the method of mathematical induction result is true $\forall n \in \mathbb{N}$

3. For $x > 0$ $\cot^{-1} x = \sin^{-1} \left(\frac{1}{\sqrt{x^2 + 1}} \right)$ and

$$\text{for } x < 0 \cot^{-1} x = \pi - \sin^{-1} \left(\frac{1}{\sqrt{x^2 + 1}} \right)$$

$$\text{In both the cases } \sin(\cot^{-1} x) = \left(\frac{1}{\sqrt{x^2 + 1}} \right)$$

$$\Rightarrow \cos \left(\tan^{-1} \frac{1}{\sqrt{1+x^2}} \right) = \cos \left(\cos^{-1} \frac{\sqrt{x^2+1}}{\sqrt{x^2+2}} \right) = \frac{\sqrt{x^2+1}}{\sqrt{x^2+2}}$$

4. E_1 : coin is fair, E_2 : coin is biased, A : first toss shows head and second toss shows tail

$$P(E_1/A) = \frac{P(A/E_1)P(E_1)}{P(A/E_1)P(E_1) + P(A/E_2)P(E_2)}$$

$$= \frac{\frac{m}{N} \cdot \frac{1}{2} \cdot \frac{1}{2}}{\frac{m}{N} \cdot \frac{1}{2} \cdot \frac{1}{2} + \frac{1-m}{N} \cdot \frac{2}{3} \cdot \frac{1}{3}} = \frac{9m}{9m + 8N - 8m} = \frac{9m}{8N + m}$$

5. The given equation can be written as $(z^p - 1)(z^q - 1) = 0$

$$\Rightarrow \alpha = e^{\frac{2k_1\pi}{p}} \text{ or } e^{\frac{2k_2\pi}{q}}$$

since p and q are prime to each other

$$\Rightarrow \text{L.C.M. of } \frac{2k_1\pi}{p} \text{ and } \frac{2k_2\pi}{q} \text{ is equal to } 2m\pi, \text{ where } m \text{ is the L.C.M. of } k_1 \text{ and } k_2 \text{ but } \alpha \neq 1.$$

Hence there is no value of α which satisfy both the equations simultaneously other than 1. So α will be the root of either $z^p - 1 = 0$ or $z^q - 1 = 0$

$$\Rightarrow \text{either } 1 + \alpha + \dots + \alpha^{p-1} = 0 \text{ or } 1 + \alpha + \dots + \alpha^{q-1} = 0$$

6. The line $y = mx$ meets the given lines in $P \left(\frac{1}{m+1}, \frac{m}{m+1} \right)$ and $Q \left(\frac{3}{m+1}, \frac{3m}{m+1} \right)$

$$\text{Hence equation of } L_1 \text{ is } y - \frac{m}{m+1} = 2 \left(x - \frac{1}{m+1} \right) \Rightarrow y - 2x - 1 = \frac{-3}{m+1} \dots \dots (1)$$

$$\text{and that of } L_2 \text{ is } y - \frac{3m}{m+1} = -3 \left(x - \frac{3}{m+1} \right) \Rightarrow y + 3x - 3 = \frac{6}{m+1} \dots \dots \dots (2)$$

$$\text{From (1) and (2) } \frac{y - 2x - 1}{y + 3x - 3} = -\frac{1}{2}$$

$$\Rightarrow x = 3y + 5 = 0 \text{ which is a straight line}$$

7. Let the equation of the line be $(y-2) = m(x-8)$ where $m < 0$

$$\Rightarrow P = \left(8 - \frac{2}{m}, 0 \right) \text{ and } Q = (0, 2 - 8m)$$

$$\text{Now } OP + OQ = \left| 8 - \frac{2}{m} \right| + |2 - 8m| = 10 + \frac{2}{-m} + 8(-m) \geq 10 + 2\sqrt{\frac{2}{-m} \times (-m)} \geq 18$$

8. Let the ellipse be $\frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$ and O be the centre

Tangent at $P(x_1, y_1)$ is $\frac{xx_1}{a^2} + \frac{yy_1}{b^2} - 1 = 0$ whose slope = $-\frac{b^2x_1}{a^2y_1}$

Focus is $S(ae, 0)$

Equation of the line perpendicular to tangent at P is $y = \frac{a^2y_1}{b^2x_1}(x - ae) \dots (1)$

Equation of OP is $y = \frac{y_1}{x_1}x \dots (2)$

(1) and (2) intersect

$$\Rightarrow \frac{y_1}{x_1}x = \frac{a^2y_1}{b^2x_1}(x - ae)$$

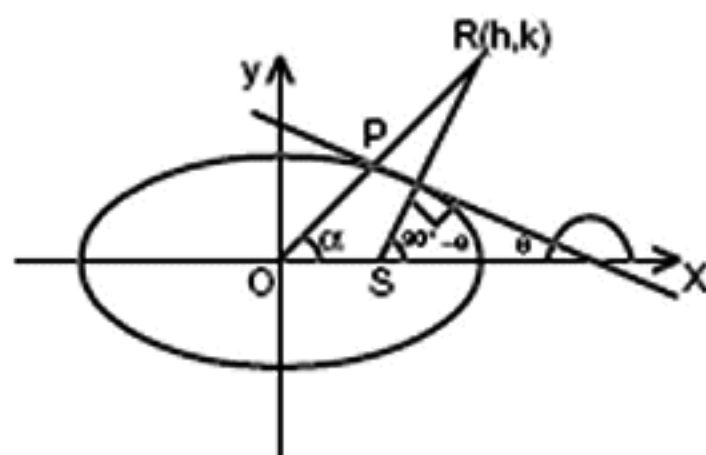
$$\Rightarrow x(a^2 - b^2) = a^3e$$

$$\Rightarrow x \cdot a^2e^2 = a^3e \Rightarrow x = \frac{a}{e}$$

which is the corresponding directrix.

Alternate

$$P = (a \cos \beta, b \sin \beta), R = (h, k)$$



$$\tan \alpha = \frac{b}{a} \tan \beta = \frac{k}{h} \dots (1)$$

and slope of tangent

$$\tan(\pi - \theta) = -\frac{b}{a} \cot \beta$$

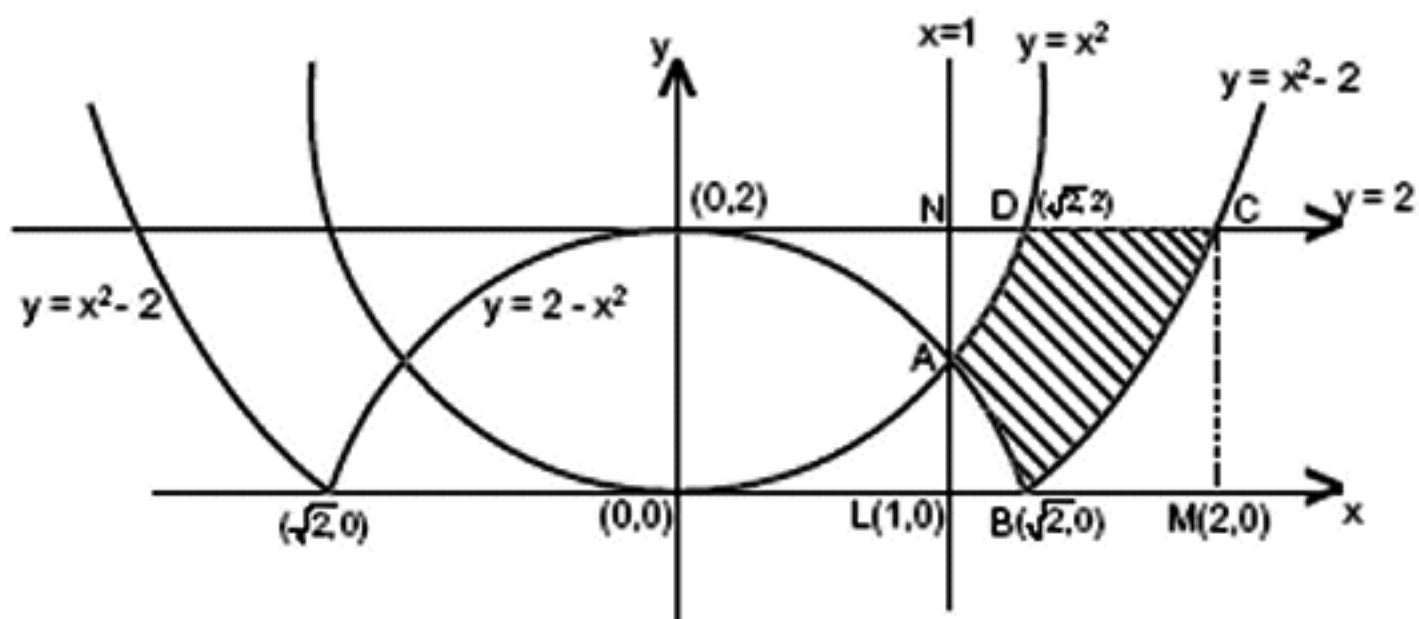
$$\therefore \tan(90^\circ - \theta) = \frac{k}{h \pm ae} = \frac{a}{b} \tan \beta = \frac{a}{b} \cdot \frac{a}{b} \cdot \frac{k}{h} \text{ (from (1))}$$

$$\text{or } \frac{k}{h \pm ae} = \frac{a^2 k}{b^2 h}$$

$$\Rightarrow h = \pm \frac{a}{e} \text{ i.e. } R \text{ lies on the corresponding directrix, when } S = (ae, 0), h = \frac{a}{e}$$

$$\text{and when } S = (-ae, 0), h = -\frac{a}{e}$$

9. Required area is ABCD



$$\int_1^{\sqrt{2}} [x^2 - (2 - x^2)] dx + 2(2 - \sqrt{2}) - \int_{\sqrt{2}}^2 (x^2 - 2) dx = \left| \frac{2x^3}{3} - 2x \right|_1^{\sqrt{2}} + 2(2 - \sqrt{2}) - \left| \frac{x^3}{3} - 2x \right|_{\sqrt{2}}^2$$

$$= \frac{2}{3}(2\sqrt{2} - 1) - 2(\sqrt{2} - 1) + 2\sqrt{2}(\sqrt{2} - 1) - \left\{ \left(\frac{8}{3} - 4 \right) - \left(\frac{2\sqrt{2}}{3} - 2\sqrt{2} \right) \right\}$$

$$= \frac{4\sqrt{2}}{3} - \frac{2}{3} + 6 - 4\sqrt{2} + \frac{4}{3} + \frac{2\sqrt{2}}{3} - 2\sqrt{2} = \frac{20}{3} + 2\sqrt{2} - 4\sqrt{2} - 2\sqrt{2} = \frac{20}{3} - 4\sqrt{2} \text{ squnits}$$

Alternate

Required area = area of rectangle LMCN - 2 (area of LBA) - area of BMC

$$= 2 - \int_1^{\sqrt{2}} (2 - x^2) dx - \int_{\sqrt{2}}^2 (x^2 - 2) dx$$

$$= \frac{20}{3} - 4\sqrt{2} \text{ squnits}$$

10. Here $V = |a \cdot (b \times c)| = \begin{vmatrix} a_1 & b_1 & c_1 \\ a_2 & b_2 & c_2 \\ a_3 & b_3 & c_3 \end{vmatrix}$

$$\text{Also } L = \frac{a_1 + b_1 + c_1 + a_2 + b_2 + c_2 + a_3 + b_3 + c_3}{3}$$

$$\geq [(a_1 + b_1 + c_1)(a_2 + b_2 + c_2)(a_3 + b_3 + c_3)]^{1/3} \quad (\because \text{AM} \geq \text{GM})$$

$$\Rightarrow L^3 \geq (a_1 + b_1 + c_1)(a_2 + b_2 + c_2)(a_3 + b_3 + c_3)$$

$$\geq |(a_1 b_2 c_3 + a_2 b_3 c_1 + a_3 b_1 c_2) - (a_1 b_3 c_2 + a_2 b_1 c_3 + a_3 b_2 c_1)| \Rightarrow L^3 \geq V.$$

Alternate

$$V = \begin{vmatrix} a_1 & b_1 & c_1 \\ a_2 & b_2 & c_2 \\ a_3 & b_3 & c_3 \end{vmatrix} = |(a_1 b_2 c_3 + a_2 b_3 c_1 + a_3 b_1 c_2) - (a_1 b_3 c_2 + a_2 b_1 c_3 + a_3 b_2 c_1)|$$

$$V \leq \max. \{(a_1 b_2 c_3 + a_2 b_3 c_1 + a_3 b_1 c_2), (a_1 b_3 c_2 + a_2 b_1 c_3 + a_3 b_2 c_1)\}$$

$$\text{Let } (a_1 b_2 c_3 + a_2 b_3 c_1 + a_3 b_1 c_2) \geq (a_1 b_3 c_2 + a_2 b_1 c_3 + a_3 b_2 c_1)$$

$$V \leq (a_1 b_2 c_3 + a_2 b_3 c_1 + a_3 b_1 c_2)$$

$$L = \left(\frac{a_1 + b_2 + c_3}{3} \right) + \left(\frac{a_2 + b_3 + c_1}{3} \right) + \left(\frac{a_3 + b_1 + c_2}{3} \right)$$

$$L \geq (a_1 b_2 c_3)^{1/3} + (a_2 b_3 c_1)^{1/3} + (a_3 b_1 c_2)^{1/3}$$

$$L^3 \geq (a_1 b_2 c_3 + a_2 b_3 c_1 + a_3 b_1 c_2) \dots \dots \dots (2)$$

From equation (1) and (2),

$$V \leq L^3.$$

similarly we can prove for the case $(a_1 b_2 c_3 + a_2 b_3 c_1 + a_3 b_1 c_2) < (a_1 b_3 c_2 + a_2 b_1 c_3 + a_3 b_2 c_1)$

11. $I = \int (x^{3m} + x^{2m} + x^m)(2x^{2m} + 3x^m + 6)^{1/m} dx, x > 0$

$$\int (x^{3m-1} + x^{2m-1} + x^{m-1})(2x^{2m} + 3x^m + 6x^m)^{1/m} dx$$

$$\text{Let } 2x^{3m} + 3x^{2m} + 6x^m = t \Rightarrow (x^{2m-1} + x^{2m-1} + x^{m-1})dx = \frac{dt}{6m}$$

$$\Rightarrow \int t^{1/m} \cdot \frac{dt}{6m} = \frac{1}{6m} \left(\frac{t^{(m+1)/m}}{\frac{m+1}{m}} \right) + c = \frac{(2x^{3m} + 3x^{2m} + 6x^m)^{(m+1)/m}}{6(m+1)} + c$$

12. $g(f(x)) = \begin{cases} x+a+1 & x < -a \\ (x+a-1)^2 + b, & -a \leq x < 0 \\ x^2 + b, & 0 \leq x < 1 \\ (x-2)^2 + b, & x \geq 1 \end{cases}$

$$\left. \begin{aligned} g(f(-a)^-) &= 1 \\ g(f(-a)^+) &= 1+b \end{aligned} \right\} \Rightarrow b=0 \text{ (as } g \circ f \text{ is continuous everywhere)}$$

$$\text{Now, } \left. \begin{aligned} g(f(0)^-) &= (a-1)^2 \\ g(f(0)^+) &= 0 \end{aligned} \right\} \Rightarrow a=1$$

For this value of a and b, g(f(x)) is continuous.

$$\text{And } g(f(x)) = \begin{cases} x+2, & x < -1 \\ x^2, & -1 \leq x < 1 \\ (x-2)^2, & x \geq 1 \end{cases}$$

Clearly at $x=0$ g(f(x)) is differentiable.